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## N-Heterocyclic Carbene Ligands in Cobalt-Catalyzed Sequential Cyclization/ Cross-Coupling Reactions of 6-Halo-1-hexene Derivatives with Grignard Reagents

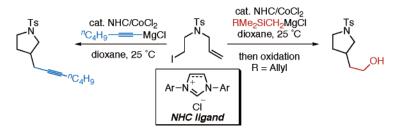
Hidenori Someya, Hirohisa Ohmiya, Hideki Yorimitsu,\* and Koichiro Oshima\*

Department of Material Chemistry, Graduate School of Engineering, Kyoto University, Kyoto-daigaku Katsura, Nishikyo-ku, Kyoto 615-8510, Japan

 $yori@orgrxn.mbox.media.kyoto-u.ac.jp;\ oshima@orgrxn.mbox.media.kyoto-u.ac.jp$ 

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## **ABSTRACT**



N-Heterocyclic carbene/cobalt systems effectively catalyze sequential cyclization/cross-coupling reactions of 6-halo-1-hexene derivatives with trialkylsilylmethyl and 1-alkynyl Grignard reagents, which phosphine and amine ligands did not promote.

N-Heterocyclic carbenes (NHCs) have been attracting increasing attention in various fields of organic chemistry. From a viewpoint of transition-metal-catalyzed reactions, they are widely used as two-electron  $\sigma$ -donor ligands for a number of reactions including carbon—carbon bond formations, carbon—heteroatom formations, and C—H activation. Among these reactions, NHC ligands display excellent performance in the palladium-catalyzed cross-coupling reactions. In contrast to the development of the NHC/palladium system, there are few examples of the use of NHC ligands in cobalt-catalyzed cross-coupling reactions. In this context, we have been exploring the possibility of NHCs in the cobalt-

catalyzed carbon—carbon bond-forming process. Herein, we present cobalt-catalyzed cross-coupling reactions of alkyl halides with Grignard reagents that involve intramolecular cyclizations,<sup>5–7</sup> which proceed only with the aid of NHC ligands (Figure 1).

In light of the importance of silyl groups as a hydroxy equivalent, NHC/cobalt-catalyzed sequential cyclization/cross-coupling reaction with allyldimethylsilylmethylmagnesium chloride was first investigated. SIEt·HCl (1, 0.025 mmol) and substrate 5 (0.5 mmol) were added to a suspension of CoCl<sub>2</sub> (0.025 mmol) in dioxane (2 mL). Then allyldimethylsilylmethylmagnesium chloride (1.5 mmol, 1 M ether solution) was added over 5 s at 25 °C. An

<sup>(1) (</sup>a) N-Heterocyclic Carbenes in Transition Metal Catalysis. In *Topics in Organometallic Chemistry*; Glorius, F., Ed.; Springer: Heidelberg, 2007; Vol. 21. (b) *N-Heterocyclic Carbenes in Synthesis*; Nolan, S. P., Ed.; Wiley-Vel: Weinheim, 2006. (c) Herrmann, W. A. *Angew. Chem., Int. Ed.* **2002**, 41, 1290—1309. (d) Bourissou, D.; Guerret, O.; Gabbaï, F. P.; Bertrand, G. *Chem. Rev.* **2000**, 100, 39—91.

<sup>(2)</sup> For reviews on the use of NHC ligands in palladium-catalyzed cross-coupling reactions, see: (a) Hillier, A. C.; Grasa, G. A.; Viciu, M. S.; Lee, H. M.; Yang, C.; Nolan, S. P. *J. Organomet. Chem.* **2002**, *653*, 69–82. (b) Littke, A. F.; Fu, G. C. *Angew. Chem., Int. Ed.* **2002**, *41*, 4176–4211.

<sup>(3)</sup> Kuno, A.; Saino, N.; Kamachi, T.; Okamoto, S. *Tetrahedron Lett.* **2006**, *47*, 2591–2594. In this report, a NHC/CoCl<sub>2</sub> system served as effectively as a Co(acac)<sub>3</sub> catalyst without any additional ligands. No significant advantage of NHC in cobalt-catalyzed cross-coupling reactions was observed.

<sup>(4)</sup> Intramolecular cyclotrimerization of triynes catalyzed by an NHC/cobalt system was reported. See: Saino, N.; Kogure, D.; Okamoto, S. *Org. Lett.* **2005**, *7*, 3065–3067.

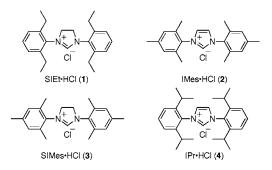


Figure 1. N-Heterocyclic carbenes.

exothermic reaction immediately took place. The mixture was stirred at 25 °C for 30 min to afford the corresponding cyclization/coupling product 6 in 81% yield (Scheme 1). The

3-(silylethyl)pyrrolidine derivative **6** underwent Tamao—Fleming oxidation to furnish the corresponding alcohol **7**.8 This reaction mechanism would consist of the following sequence:<sup>6</sup> (1) generation of the corresponding carbon-

(5) For recent reports on cobalt-catalyzed coupling reactions, see: (a) Cahiez, G.; Avedissian, H. *Tetrahedron Lett.* **1998**, *39*, 6159–6162. (b) Avedissian, H.; Bérillon, L.; Cahiez, G.; Knochel, P. *Tetrahedron Lett.* **1998**, *39*, 6163–6166. (c) Nishii, Y.; Wakasugi, K.; Tanabe, Y. *Synlett* **1998**, 67–69. (d) Korn, T. J.; Knochel, P. *Angew. Chem., Int. Ed.* **2005**, *44*, 2947–2951. (e) Gomes, P.; Gosmini, C.; Périchon, J. *Org. Lett.* **2003**, *5*, 1043–1045. (f) Amatore, M.; Gosmini, C.; Périchon, J. *Eur. J. Org. Chem.* **2005**, 989–992.

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(7) Several groups reported the use of NHC ligands in the palladium-and iron-catalyzed cross-coupling reactions of alkyl halides with organometallic reagents. See: (a) Frisch, A. C.; Rataboul, F.; Zapf, A.; Beller, M. J. Organomet. Chem. 2003, 687, 403–409. (b) Eckhardt, M.; Fu, G. C. J. Am. Chem. Soc. 2003, 125, 13642–13643. (c) Arentsen, K.; Caddick, S.; Cloke, F. G. N.; Herring, A. P.; Hitchcock, P. B. Tetrahedron Lett. 2004, 45, 3511–3515. (d) Hadei, N.; Kantchev, E. A. B.; O'Brien, C. J.; Organ, M. G. Org. Lett. 2005, 7, 3805–3807. (e) Bedford, R. B.; Betham, M.; Bruce, D. W.; Danopoulos, A. A.; Frost, R. M.; Hird, M. J. Org. Chem. 2006, 71, 1104–1110. (f) Altenhoff, G.; Würtz, S.; Glorius, F. Tetrahedron Lett. 2006, 47, 2925–2928.

(8) (a) Tamao, K.; Nakajima, T.; Kumada, M. *Organometallics* **1984**, 3, 1655–1660. (b) Fleming, I.; Henning, R.; Plaut, H. *J. Chem. Soc., Chem. Commun.* **1984**, 29–31. (c) Klos, A. M.; Heintzelman, G. R.; Weinreb, S. M. *J. Org. Chem.* **1997**, 62, 3758–3761. For reviews on carbon–silicon bond oxidation, see: (d) Jones, G. R.; Landais, Y. *Tetrahedron* **1996**, 52, 7599–7662

centered radical from **5** by single-electron transfer from an electron-rich cobalt complex, (2) radical cyclization, (3) capture of the 3-pyrrolidinomethyl radical by a cobalt complex, and (4) reductive elimination.

A variety of imidazolium salts were screened to reveal that SIEt·HCl (1) was the best ligand. For example, the use of 1,3-dialkyl-substituted imidazolium salts (not shown in Figure 1) resulted in no more than 5% yields of 6, and significant amounts of 1-toluenesulfonyl-3-methylenepyrrolidine were instead formed via  $\beta$ -elimination. On the other hand, 1,3-dimesityl-substituted derivative 2 showed modest activity (36% yield), and the use of 3, the dihydro analogue of 2, further improved the yield of 6 (54% yield). IPr•HCl (4) that bears larger aryl groups than IMes·HCl (2) furnished none of the coupling product, leaving most of the starting material. The use of other ligands such as phosphines (PPh<sub>3</sub>, P<sup>r</sup>Bu<sub>3</sub>, and 1,3-bis(diphenylphosphino)propane) and diamines (N,N,N',N'-tetramethylethylenediamine) and N,N,N',N'-tetramethyl-1,2-cyclohexanediamine<sup>6g</sup>) resulted in much lower yields (<10%).

Various substrates were examined,<sup>9</sup> and the results are listed in Table 1. Halo acetals bearing a terminal alkene

**Table 1.** Representative Examples of NHC/Cobalt-Catalyzed Sequential Cyclization/Cross-Coupling Reaction<sup>a</sup>

entry	substrate		product (R = Allyl)	yield %	6
1	0,,0	8		1; SiMe <sub>2</sub> R	<b>3</b> 72 (85/15)
<sup>n</sup> C₄l 2	19070	9	nC4H9O	< 14 _SiMe₂R	81 (67/33)
3 <sup>n</sup> C <sub>4</sub> i	H <sub>9</sub> O <sub>3</sub> O MC <sub>5</sub> H <sub>11</sub>	10 <sup>b</sup>	"C <sub>4</sub> H <sub>9</sub> O <sub>4</sub>	<b>-</b> <sup>n</sup> C <sub>5</sub> H <sub>11</sub> <b>15</b> SiMe₂R	78 (54/46)
4 5	×	11 (X 12 (X	_/	16 1e <sub>2</sub> R	67 (X = I) 18 (X =Br)
6°	Ts N	5	Ts N S	<b>17</b> iMe <sub>2</sub> Ph	. 78

 $^a$  The reaction conditions are described in Scheme 1 (5→6).  $^b$  1:1 mixture of diastereomers.  $^c$  Dimethylphenylsilylmethylmagnesium chloride was employed.

moiety underwent the cyclization/coupling reactions to give the corresponding silylethyl-substituted tetrahydrofuran derivatives in good yields (entries 1–3). Carbocycle **16** was obtained exclusively in the reaction of 6-iodo-1-hexene (**11**) in 67% yield (entry 4). The corresponding bromide **12** was

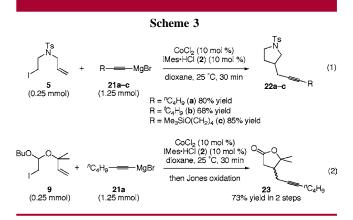
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<sup>(9)</sup> The NHC/cobalt system could be employed for cross-coupling reactions of primary alkyl halides without a cyclization process. For instance, treatment of isobutyl iodide (0.5 mmol) with allyldimethylsilylmethylmagnesium chloride (1.5 mmol, 1 M ether solution) in dioxane (2 mL) in the presence of SIEt+HCl (1, 0.025 mmol) and CoCl $_2$  (0.025 mmol) for 30 min at 25 °C afforded the corresponding coupling product in 79% yield.

less reactive (entry 5). Dimethylphenylsilylmethylmagnesium chloride was also available for this reaction (entry 6).

The silicon-tethered 6-iodo-1-hexene derivative **18** was employed for the NHC/cobalt-catalyzed reaction, and the corresponding cyclization/coupling product **19** could be easily transformed to the diol **20** after oxidation with alkaline hydrogen peroxide (Scheme 2).<sup>10</sup>

We have recently reported TMEDA/cobalt-mediated alkynylation of alkyl halides.6h However, the scope of the alkynyl Grignard reagents available for use is limited. Namely, we obtained promising results only with 2-trimethylsilylethynylmagnesium bromide. The previous conditions could not be applied to alkyl-substituted alkynyl Grignard reagents such as 1-hexynylmagnesium bromide. On the other hand, NHC ligand 2 proved to be effective for sequential cyclization/cross-coupling reactions with 1-alkynyl Grignard reagents (Scheme 3). For example, treatment of 5 with 1-hexynylmagnesium bromide (21a) in the presence of IMes·HCl (2) and CoCl<sub>2</sub> provided alkynylated product 22a in 80% yield (eq 1). The alkynes 21b and 21c, bearing a sterically bulky group and a siloxy group, respectively, reacted smoothly. This alkynylation required the 1,3-dimesityl-substituted imidazolium salt 2 to attain satisfactory results.



The reactions with the aid of SIEt·HCl (1), SIMes·HCl (3), and IPr·HCl (4) gave none of the coupling product and resulted in recovery of the starting material. Treatment of 9 led to a highly effective cyclization/alkynylation followed by Jones oxidation producing lactone 23 in 73% yield (eq 2).

In summary, NHC ligands **1** and **2** have emerged as irreplaceable ligands in fast-growing cobalt-catalyzed cross-coupling reactions.<sup>6</sup> With the aid of the NHC ligands, we have developed new and useful variants of sequential cyclization/coupling reactions of 6-halo-1-hexene derivatives with trialkylsilylmethyl and 1-alkynyl Grignard reagents.

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**Supporting Information Available:** Experimental details and characterization data for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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